

# Synthesis of 2-Isloxazolines From Olefins Derived From Norephedrine And Pulegone

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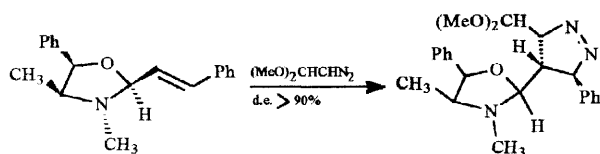
**Abstract:** In this communication we report the use of ( $\pm$ )-norephedrine and (-)-8-benzylaminomenthol (derived from (+)-pulegone) as chiral adjuvants for the cycloaddition of  $\alpha,\beta$ -unsaturated 1,3-dipolarophiles. 2-Isloxazolines were obtained with low stereoselectivity from the reaction of nitrile oxides with the N-tosyl norephedrine derivative as the dipolarophile. Cycloaddition of nitrile oxide with 2-vinyl-N-benzyl-4,4,7 $\alpha$ -trimethyl-*trans*-octahydro-1,3-benzoxazine produced stereoisomeric 2-isloxazolines in a ratio of about 95:5.

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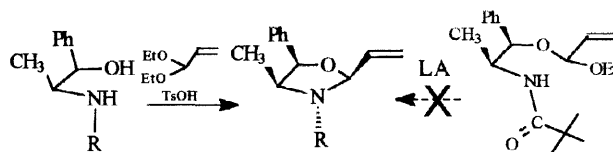
*Keywords:* amino alcohols; asymmetric induction; cycloadditions; isloxazolines

Substituted 2-isloxazolines are an important class of heterocycle, as they can be selectively transformed into 1,3-amino alcohols or  $\beta$ -hydroxy ketones which are useful intermediates in the total synthesis of a wide variety of natural products.<sup>1</sup> It is known that monosubstituted olefins exhibit high regioselectivity in 1,3-dipolar cycloadditions with nitrile oxides to give 5-substituted 2-isloxazolines as the major product.<sup>1</sup> Further stereoselective cycloadditions with a high level of asymmetric induction have been reported by Ukaji, Curran, Oppolzer, Kim, and Olsson.<sup>2</sup> Nonetheless, a mixture of 2-isloxazolines regioisomers generally results if unsymmetrical 1,2-disubstituted olefins including  $\alpha,\beta$ -unsaturated esters and ketones are used. The diastereoselectivity in these 1,3-dipolar cycloadditions is also much lower.<sup>1,2b,c,d</sup> Kanemassa and Ukaji have reported better regio- and stereoselectivity in the 1,3-dipolar cycloaddition involving magnesium crotyl alcoholate<sup>3a</sup> and allylic alcohol<sup>3b</sup> by the coordination of benzonitrile oxide with magnesium or zinc ions. The use of  $\alpha,\beta$ -unsaturated aldehydes is restricted due to the formation of a *bis*-cycloadduct resulting from the addition of nitrile oxide to the carbonyl group of the initial 2-isloxazoline product.<sup>4a</sup> This problem is alleviated by the use of  $\beta$ -substituted  $\alpha,\beta$ -unsaturated aldehyde equivalents (i.e. acetals and thioacetals). Interestingly, these dipolarophiles are highly regioselective and afford predominantly 2-isloxazolines bearing an acetal group at the C-4 position<sup>4b</sup> or a 2-isloxazoline with a dithiane group at the C-5 position.<sup>4c</sup> In this communication, we report the use of ( $\pm$ )-norephedrine and (+)-pulegone as carbonyl protecting groups. This choice fulfills the following requirements: commercial availability of both

enantiomeric forms of the chiral auxiliary and a high degree of  $\Pi$ -face differentiation in other asymmetric processes.<sup>5</sup> For example, ephedrine has been used as a chiral adjuvant to cinnamaldehyde to induce a high level of asymmetry during a 1,3-dipolar cycloaddition with 1,1-dimethoxy-2-diazomethane (**Scheme 1**).<sup>5a</sup> Thus, it was anticipated that the cycloaddition of 2-alkenyl oxazolidine (derived from ( $\pm$ )-norephedrine) using various nitrile oxides as the dipole should proceed with equal facial selectivity.



Scheme 1

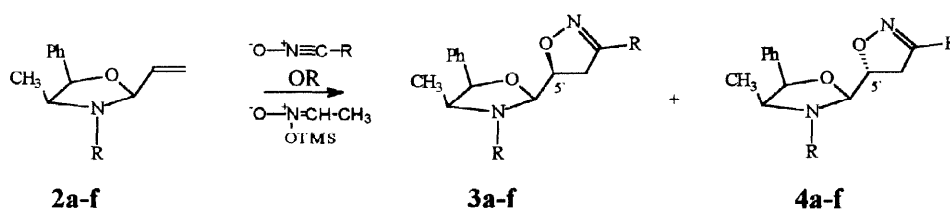


Scheme 2 1a-f

2a-f

5

N-Substituted oxazolidines **2a-f** were prepared from the corresponding N-substituted ( $\pm$ )-norephedrines **1a-f** following literature procedures (**Scheme 2**).<sup>6</sup> In the case of the N-pivaloyl derivative **1e**, a diastereomeric mixture of the uncyclized **5** was obtained in 89% yield.<sup>7</sup> The N-pivaloyl oxazoline derivative **2e** was obtained in low yield from the condensation of ( $\pm$ )-norephedrine with acrolein followed by the addition of pivaloyl chloride to the oxazoline formed *in situ*.<sup>8</sup> Olefins **2a-f** were treated with acetonitrile oxide or benzonitrile oxide generated *in situ* under various reaction conditions. The results are summarized in Table 1.



2a-f

3a-f

4a-f

Table 1 1,3-Dipolar Cycloaddition of **2a-f** with acetonitrile oxide and phenylnitrile oxide.

No.	Dipolarophile <b>2a-f</b>	Dipole Method (eq)	Additive (eq)	Yield <sup>d</sup> (%)	Adduct Ratio (%) <sup>e</sup> ( <b>3a-f</b> : <b>4a-f</b> )
Acetonitrile oxide					
1	<b>2a</b> , R=CH <sub>3</sub>	b (10)	-	-	-
2	<b>2b</b> , R=COCH <sub>3</sub>	a;b;c (2; 5; 2) -	-	-	-
3	<b>2c</b> , R=CO <sub>2</sub> CH <sub>2</sub> CF <sub>3</sub>	b (1)	-	28	1 : 1
4	<b>2d</b> , R=CO <sub>2</sub> <i>i</i> Pr	b (10)	-	45	1 : 1
5	<b>2e</b> , R=CO <sub>2</sub> <i>t</i> -Bu	b (10)	-	57	1 : 1
6	<b>2f</b> , R=Ts	a (1.2)	-	31	1 : 1
7	<b>2f</b> , R=Ts	b (10)	-	50	1.5 : 1
8	<b>2f</b> , R=Ts	b (10)	MgBr <sub>2</sub> •Et <sub>2</sub> O(10eq) <sup>9b</sup>	29	1.3 : 1
9	<b>2f</b> , R=Ts	c (6)	MgBr <sub>2</sub> •Et <sub>2</sub> O(1eq)	22	1.4 : 1
10	<b>2f</b> , R=Ts	c (5)	ZnCl <sub>2</sub> (6eq)	11	1 : 1
Phenylnitrile oxide					
11	<b>2f</b> , R=Ts	b (3)	-	52	1.1 : 1
12	<b>2f</b> , R=Ts	b <sup>3a</sup> (10)	MgBrCl(10eq) <sup>9a</sup>	15	1 : 1

a) To a solution of **2a-f** in benzene at RT was slowly added nitroethane, phenylisocyanate and Et<sub>3</sub>N in catalytic amounts.

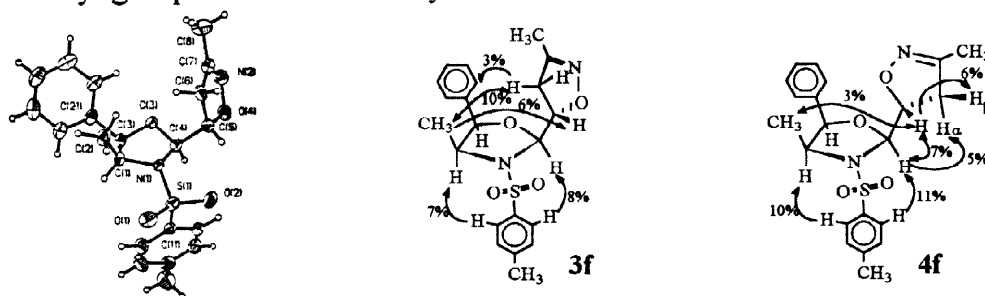
b) To a solution of RCCI=NOH was slowly added a stoichiometric amount of Et<sub>3</sub>N, **2a-f** (1eq) in CH<sub>2</sub>Cl<sub>2</sub> and the Lewis acid.

c) To a solution of **2a-f** (1eq) in CH<sub>2</sub>Cl<sub>2</sub> was added the trimethylsilyl ester of *aci*-nitroetane and the Lewis acid.

For a, b, and c, the reagents were added at 0°C, followed by stirring overnight at room temperature.

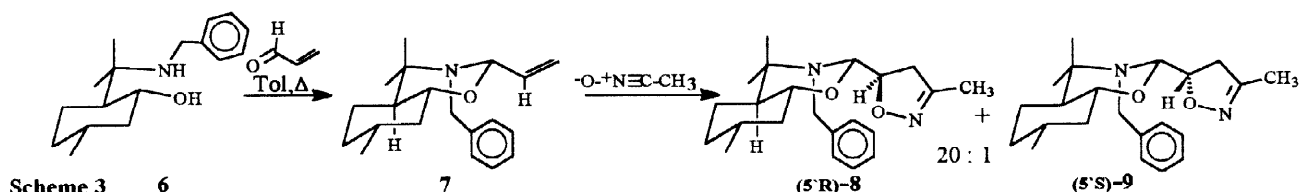
d) Isolated yield. e) Determined by <sup>1</sup>H NMR and <sup>13</sup>C analysis of the crude reaction mixture.

While the N-methyl and N-acetyl derivatives were found to be unsuitable using the reaction conditions required to generate the nitrile oxide (i.e. by method a, b or c), the carbamate and N-tosyl derivatives led to a mixture of the two epimers at C-5' (i.e. **3a-f** and **4a-f**).<sup>10,11</sup> Method b, involving the base conversion of excess  $\alpha$ -chloro oxime into nitrile oxide gave higher conversion yields. We observed no asymmetric induction for the carbamate substituent (entry 3-5), while the N-tosyl group gave **3f** with 20% *de* (entry 7-10). Unfortunately, the addition of a Lewis acid had no effect other than to decrease the chemical yield.<sup>9</sup> The stereochemistry of **3f** was unambiguously determined by X-ray diffraction analysis, and in solution, the results of nOe experiments for **3f** and **4f** were consistent with the relative stereochemistry as depicted in **Figure 1**. From inspection of **Figure 1**, we attributed the low level of asymmetric induction to the *trans* relationship between the N-tosyl and the C-2 vinyl groups. This leaves the face of the olefin opposite to the tosyl group accessible for a cycloaddition for each reactive conformer.



**Figure 1** Left: X-ray crystal structure of isoxazoline **3f**. Right: Relevant nOe signals in **3f** and **4f**.

Having demonstrated that oxazolidines can serve as chiral adjvants for  $\alpha,\beta$ -unsaturated 1,3-dipolarophiles we decided to use a more conformationally rigid auxiliary such as the *trans*-fused octahydrobenzoxazine **7**, derived from (+)-pulegone. Eliel's<sup>12a</sup> procedure for the preparation of (-)-8'-benzylaminomenthol **6**, was used followed by the condensation of **6** with acrolein to afford **7**.<sup>12b</sup> The conformation of the equatorial C-2 vinyl group and the axial N-benzyl residue was confirmed by nOe experiments.<sup>13</sup> The 1,3-dipolar cycloaddition of acetonitrile oxide with **7** gave a mixture of diastereomers **8** and **9**, and was found to be highly stereoselective (i.e., 90% *de* in favor of **8**, see **Scheme 3**). The absolute stereochemistry at the C-5' center for the epimeric

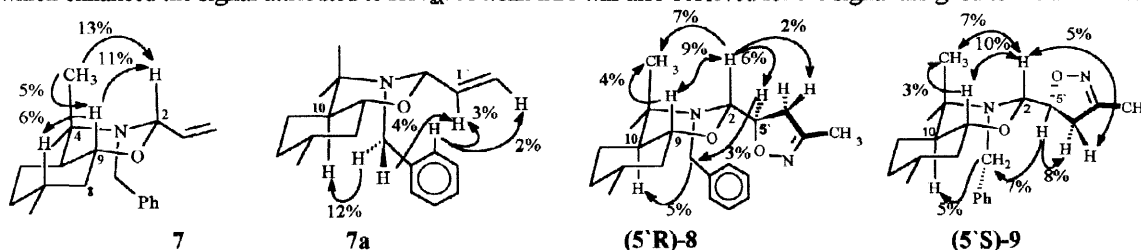


2-isoxazolines **8** and **9** is supported by nOe experiments.<sup>13</sup> Assuming that the reactive conformation in the cycloaddition is close to that highly favored in the ground state, the (5'R) stereochemistry of the major epimer **8** is readily explained by the cycloaddition of the nitrile oxide on the *Re*-face of the olefin, **7**, opposite to the N-benzyl substituent.

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- The intermediate **5** shows an absorption band at 3286  $\text{cm}^{-1}$  in the IR and two resonances at  $\delta$  6.25 (s) and 6.28 (s) in the  $^1\text{H}$  NMR spectrum for the free N-H function. Attempts to cyclize **5** to its corresponding isoxazoline, **2e**, under acidic conditions led to decomposition (i.e. PPTs, TsOH,  $\text{BF}_3 \cdot \text{Et}_2\text{O}$ ).
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- a) While we found an acceleration and a good regiocontrol for the cycloaddition of  $\text{Ph-C}\equiv\text{N}^+\text{-O}^-\text{MgBr}$  to crotyl magnesium alcoholate, the use of  $\text{Ph-C}\equiv\text{N}^+\text{-O}^-\text{MgBr}$  as described by Kamenasa<sup>3a</sup> had no effect on product distribution and a low conversion yield was observed. b) In the case of the reaction with  $\text{CH}_3\text{-C}\equiv\text{N}^+\text{-O}^-\text{MgBr}$  generated under the same conditions no cycloaddition took place.
- All new compounds gave the correct high resolution mass spectra and suitable spectroscopic data (ir, nmr, ms). The crystallographic structure for **3f** will be published in *Acta Crystallographica*.
- Spectroscopic data for isoxazoline **4f**:  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  7.84 (d, 2H,  $J=8.2$ , Ar), 7.39 (d, 2H,  $J=8.5$ , Ar), 7.27 (m, 3H, Ar), 7.13 (dd, 2H,  $J=1.7$ ,  $J=7.8$ , Ar), 5.16 (d, 1H,  $J=3.3$ , H-2), 4.88 (dq, 1H,  $J_{5-2}=3.7$ ,  $J_{5-4\alpha}=6.6$ ,  $J_{5-4\beta}=10.6$  H-5'), 4.34 (d, 1H,  $J=5.5$ , H-5), 4.16 (t, 1H,  $J=6.6$ ,  $J=6.0$ , H-4), 3.23 (dd, 1H,  $J_{4\alpha 5}=6.4$ ,  $J=17.5$ , H-4'  $\alpha$ ), 3.13 (dd, 1H,  $J_{4\beta 5}=10.8$ ,  $J=17.2$ , H-4'  $\beta$ ), 2.47 (s, 3H, Ph- $\text{CH}_3$ ), 2.07 (s, 3H,  $\text{CH}_3\text{-3}'$ ), 0.86 (d, 3H,  $J=6.8$ ,  $\text{CH}_3\text{-4}$ );  $^{13}\text{C}$  NMR:  $\delta$  130.1 (Ar), 128.1 (Ar), 128.3 (Ar), 127.9 (Ar), 126.1 (Ar), 155.4 (Ar), 144.6 (Ar), 135.1 (Ar), 134.9 (Ar), 91.1 (C2), 81.8 (C5), 80.4 (C5'), 58.7 (CH-4), 40.5 (C4'), 21.7 ( $\text{CH}_3\text{-Ar}$ ), 16.7 ( $\text{CH}_3\text{-3}'$ ), 13.0 ( $\text{CH}_3\text{-4}$ ).
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- Eliel and Pedrosa first reported an equatorial [12a,c,e] and later an axial[12b,d] stereochemistry for the N-benzyl group in 2-alkyl perhydrobenzoxazines similar to **7**. We found that the irradiation of benzylic protons in **7**, **8** and **9** caused a strong nOe which enhanced the signal attributed to H10<sub>ax</sub>. A weak nOe was also observed for the signal assigned to the axial methyl at C4.



Based on these results, we concluded that the N-benzyl substituent is located mainly in the axial position. An axial position for proton H-2 in compounds **7**, **8** and **9** is supported by the observation of nOe enhancement of the signals attributed to H-9<sub>ax</sub> and the axial methyl at C-4. The C-2 vinyl group is therefore in the equatorial position. The assignment of resonances for H-2, **8**, **9** and **10** was supported by COSY and HMQC cross peak correlations.